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LIGNIN AND RELATED COMPOUNDS X. THE SYNTHESIS OF DIMERIC-TYPE COMPOUNDS DERIVED FROM LIGNIN

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ABSTRACT

Four dimeric-type compounds (14, 15, 16, and 17) that had been isolated earlier from the catalytic hydrogenolysis products of aspen poplar lignin have been synthesized for structural verification.

INTRODUCTION

In a previous communication¹ evidence was presented in the form of NMR and MS data for the characterization of the dimerictype compounds, <u>14</u>, <u>15</u>, <u>16</u>, and <u>17</u>, as products of the catalytic (Rh-C) hydrogenolysis of aspen poplar lignin. This paper reports the syntheses and characterization of these compounds as a final verification of their structure. The synthesis involves the base-catalyzed condensation of two appropriately substituted monomers, one a phenylmethyl chloride and the other an ethyl or methyl phenylethanoate. The appropriate starting materials, <u>1</u>, <u>2</u>, <u>3</u>, <u>4</u>, and <u>5</u> were prepared as follows and identified by comparison with reported analytical data. Conversion of vanillin (4-hydroxy-3-methoxybenzaldehyde) or syringaldehyde (4-hydroxy-3, 5-dimethoxybenzaldehyde) into the corresponding benzyl ethers²,³

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and reduction with sodium borohydride gave the substituted benzyl alcohols^{4,5}. Subsequent treatments of these alcohols with thionyl chloride in chloroform gave the required substituted benzyl chlorides, $\underline{1}^6$ and $\underline{2}^{5,6}$. Further treatment of the chlorides with sodium cyanide gave the nitriles^{5,6} which on hydrolysis yielded the ethanoic acids^{5,7,8}. Esterification of the acids gave the required but previously unreported ethyl esters 3 and 4 and the reported methyl ester $\underline{5}^9$. Overall yields starting from vanillin and syringaldehyde were 59 and 58 percent for 1 and 2, respectively, 26 percent for $\underline{3}/\underline{5}$ and 25 percent for $\underline{4}$.

An alternative method for the preparation of the homoacids from acetoguaiacone (1-(4-hydroxy-3-methoxyphenyl)ethanone) and acetosyringone(1-(4-hydroxy-3,5-dimethoxyphenyl)ethanone) using the Willgerodt-Kindler reaction¹⁰ was less satisfactory.

The dimerization reaction, catalyzed by lithium N-isopropylcyclohexylamide¹¹, and the subsequent conversion to the required hydrogenolysis products was effected as shown in Scheme 1.

EXPERIMENTAL

A. Apparatus

¹E nmr spectra were recorded on either a Varian T-60 and/or EM-90 instrument with TMS as internal standard.

Mass spectra were obtained using an MS-12 mass spectrometer. Column chromatographic separations were made using a column (60 cm x 2.8 cm) packed with Bio-Sil A (silica gel, 200 - 400 mesh). Thin layer chromatographic separations were made using Merck TLC aluminum sheets (20 x 20 cm) precoated with silica gel 60F₂₅₄.

B. Synthesis of Monomeric Starting Materials

<u>Ethyl 4-benzyloxy-3-methoxyphenylethanoate</u>, 3, was prepared as a yellow oil by the esterification of the appropriately substituted ethanoic acid according to the method of Kametani <u>et</u> al⁵. Yield, 84%. Calcd. for $C_{18} H_{20} O_4$: C, 71.98; H, 6.71%.



Found: C, 71.75, H, 6.86%. ¹H nmr (CDCl₃) δ ppm 1.15 (t, 3H, CH₃, J = 6.5 Hz), 3.6 (s, 2H, CH₂), 3.9 (s, 3H, OCH₃), 4.3 (q, 2H, OCH₂CH₃, J = 6.5 Hz), 5.2 (s, 2H, PhCH₂O), 6.8-6.9 (m, 3H, Ph), 7.2 - 7.6 (m, 5H, ArCH₂). Mass spec. (70 ev) m/z 300 (M⁺, 15%), 272 (31%), 227 (40%), 181 (10%) and 91 (100%).

Ethyl 4-benzyloxy-3,5-dimethoxyphenylethanoate, 4, was prepared from the appropriate starting materials as above. Yield, 82%. Calcd. for $C_{19}H_{22}O_5$: C, 69.07; H, 6.71%. Found: C, 69.18; H, 6.60%. ¹H nmr (acetone-d₆) δ ppm 1.2 (t, 3H, CH₃, J = 6.5 Hz), 3.5 (s, 2H, CH₂), 3.9 (s, 6H, OCH₃), 4.1 (q, 2H, OCH₂CH₃, J = 6.5 Hz), 4.7 (s, 2H, PhCH₂O), 6.6 (s, 2H, Ar-CH₂), 7.1 - 7.5 (m, 5H, Ph). Mass spec. (70 ev) m/z 330 (H^{+*}16%), 302 (30%), 257 (45%), 211 (12%) and 91 (100%).

C. Synthesis of Dimeric Hydrogenolysis Products

The typical procedure used to effect the synthesis of the dimeric products is illustrated below by the condensation of $\underline{1}$ and $\underline{3}$ to produce $\underline{6}$ and its subsequent conversion through $\underline{10}$ to the required $\underline{14}$. The other dimers were prepared by similar procedures.

<u>Bthyl 2,3-di(4-benzyloxy-3-methoxyphenyl) propanoate</u>, <u>6</u>.

To a magnetically stirred solution of N-isopropylcyclohexylamide (0.75g ~ 5.5 mmol) in dry THF (7.5 mL) contained in a 50 mL pear-shaped flask and cooled in a dry ice-acetone bath, was added, by syringe, 3.5 mL of 1.65M n-butyllithium (5.5 mmol) in To the stirred cold solution was added, by syringe, hexane. ethyl 4-benzyloxy-3-methoxyphenyl ethanoate (1.5g ~ 5 mmol) in dry THF (10 mL) and stirring continued for 15 minutes. After varming to room temperature this solution was added, by syringe, over a period of several minutes, to a stirred solution of 4-benzyloxy-3-methoxyphenylmethyl chloride (1.6g ~ 6.1 mmol) in dry DMSO (5 mL) in a round-bottomed flask (100 mL). After stirring at room temperature under nitrogen overnight, the resulting light brown solution which contained a small amount of solid was diluted with water (50 mL) and extracted with ether

(4 x 25 mL). The combined ether extract was washed with water (4 x 25 mL), hydrochloric acid (10%, 3 x 25 mL), saturated sodium bicarbonate solution (2 x 25 mL), saturated sodium chloride (2 x 25 mL), water (2 x 25 mL) and finally dried over anhydrous sodium sulfate. After filtration the solvent was evaporated to leave a viscous yellow-brown oil (2.8g). Purification by column chromatography using as eluant increasingly polar mixtures of ethyl ethanoate-Skelly B of v/v 1/9, 2/8, 3/7 gave several fractions from which those were combined that contained 15 as monitored by thin-layer chromatography. Evaporation of the solvents gave a crude crystalline compound which was recrystallized from cyclohexane. Yield 1.34g, 51%, m.p. 105-106°C. Calcd. for C33H3406: C, 75.26; H, 6.51%. Found: C, 75.16; H, 6.44%. ¹H nmr (CDCl₃), δ ppm 1.12(t, 3H, CH₃, J = 7.0 Hz), 2.90(q, 1H, α Ha, J = 6.8 Hz, J = 13.6 Hz), 3.33(q, 1H, α Hb, J = 8.6 Hz, J = 13.6 Hz), $3.68(q, 1H, \beta H, J = 6.8$ Hz, J = 8.6 Hz), 3.76 (s, 3E, OCH₂), 3.84 (s, 3H, OCH₂), 4.06 (q, 2H, OCH₂CH₃, J = 7.0 Hz), 5.09 (s, 2H, OCH₂Ph), 5.11 (s, 2H, OCH₂Ph), 6.56 - 6.84 (m, 6H, Ar), 7.24 - 7.42 (m, 10H, PhCH₂). Mass Spec. (70ev), m/m 526 $(\texttt{H}^+,\texttt{5X})$, m/z = 453 (1.1X), 452 (0.55X), 299 (5X), 227 (62X), 91 (100X), 65 (16X), 107 (8X).

2,3-Di(4-benzyloxy-3-methoxyphenyl)-1-propanol, 10

To a magnetically-stirred solution of lithium aluminum hydride (0.13g $^{-}$ 3.43 mmol) in diethyl ether (10 mL) was added, by syringe, a solution of <u>6</u> (0.40g $^{-}$ 0.73 mmol in 25 mL dry ether). The resulting mixture was heated under reflux with stirring for 3 hours. After cooling, ether (10 mL) was added followed by water (50 mL) and hydrochloric acid (10%, 50 mL) and finally extracted with ether (4 x 25 mL). The combined ether extract was washed with water (2 x 25 mL), saturated sodium bicarbonate solution (2 x 25 mL), saturated sodium chloride solution (2 x 25 mL), water (2 x 25 mL) and dried over anhydrous magnesium sulfate. After filtration the solvent was evaporated to give a sticky white solid <u>10</u> (0.3g, yield 81%), m.p. 90 - 92^oC (from benzene-cyclohexane). Calcd. for $C_{31}H_{32}O_5$: C, 76.84; H, 6.66%. Found: C, 76.62; H, 6.83%. ¹H nmr, (CDCl₃), δ ppm, 1.48(s, broad, 1H, OH), 2.67 - 2.89 (m, 3H, α H and β H), 3.58 -3.68 (2H, γ H overlapped with OCH₃), 3.65 (s, 3H, OCH₃), 3.73 (s, 3H, OCH₃), 4.99 (s, 2H, OCH₂Ph), 5.02 (s, 2H, OCH₂Ph), 6.48 -6.70 (m, 6H, Ar), 7.18 - 7.35 (m, 10H, PhCH₂). Mass spec. (70 ev), m/z 484 (H^{+*}, 9%), 257 (32%), 227 (35%), 240 (6%), 107 (10%), 91 (100%), 65 (25%).

2,3-Di(4-hydroxy-3-methoxyphexyl)-1-propenol, 14

2,3-Di(4-benzyloxy-3-methoxyphenyl)-1-propanol <u>10</u> (0.12g) vas dissolved in dioxane:acetic acid:water (2:2:1 v/v, 15 mL), Pd-black (0.02g) was added and the mixture stirred in the presence of hydrogen gas for 3 hours. After filtration the filtrate was extracted with chloroform (3 x 15 mL), the extract was washed with water (2 x 10 mL), dried over anhydrous sodium sulphate and evaporated to leave an oily residue. The oily mass was purified by micro pipette column packed with Bio-Sil silica gel. Calcd. for $C_{17}H_{20}O_5$: C, 67.10; H, 6.57%. Found: C, 67.29; H, 6.42%. ¹H mmr(CDCl₃, D₂O) & ppm 2.70 - 2.90 (m, 3H, α H and β H), 3.60 - 3.69 (m, 2H, γ H overlapped with OCH₃), 3.70 (s, 3H, OCH₃), 3.75 (s, 3H, OCH₃), 6.50 - 7.40 (m, 6H, Ar). Mass spec. (70 ev), m/z 304 (M⁺⁺, 16%), 150 (41%), 167 (100%), 137 (88%). Methyl 2-(4-benzyloxy-3-methoxyphenyl)-3-(4-benzyloxy-3,5dimethyoxyphenyl) propanoate, 7

It was prepared as described for compound <u>6</u>, using methyl 4-benzyloxy-3-methoxyphenyl ethanoate (1.43g $\tilde{}$ 5 mmol) and 4-benzyloxy-3,5-dimethoxyphenyl chloride (1.46g $\tilde{}$ 5 mmol). Yield 1.27g $\tilde{}$ 47%, m.p. 106 - 108°C (Skelly B:ethyl ethanoate). Calcd. for C₃₃H₃₄O₇: C, 73.04; H, 6.32%. Found: C, 72.85; H, 6.08. ¹H nmr (CDCl₃), δ ppm 2.85 (q, 1H, α Ha, J = 7.0 Hz, J $\tilde{}$ 13.6 Hz), 3.25 (q, 1H, α Hb, J = 8.4 Hz), 3.52 - 3.63 (m, 1H, β H, overlapped with OCH₃), 3.52 (s, 3H, COOCH₃), 3.62 (s, 6H, OCH₃), 3.77 (s, 3H, OCH₃), 4.87 (s, 2H, OCH₂Ph), 5.03 (s, 2H, OCH₂Ph), 6.18 - 6.90 (m, 5H, Ar), 7.24 - 7.32 (m, 10H, OCH₂Ph). Mass spec. (70 ev), m/z 542 (H⁺, 5%), 483 (0.65%), 482 (0.4%), 285
(62%), 257 (23%), 107 (8%), 91 (100%), 65 (16%).
Ethyl 2-(4-benzyloxy-3,5-dimethoxyphenyl)-3-(4-benzyloxy-3-methoxyphenyl) propanoate, 8

It was prepared as described for compound <u>6</u>, using ethyl 4-benzyloxy-3,5-dimethoxyphenyl ethanoate (1.65g $\tilde{}$ 5 mmol) and 4-benzyloxy-3-methoxybenzylchloride (1.57g $\tilde{}$ 5 mmol). Yield 1.32g $\tilde{}$ 48%, m.p. 110 - 112°C (Skelly B:ethyl ethanoate). Calcd. for C₃₄H₃₆O₇: C, 73.36; H, 6.52%. Found: C, 73.59; H, 6.18%. ¹H nmr, (CDCl₃), δ ppm 1.10 (t, 3H, CH₃, J = 7.2 Hz), 2.91 (q, 1H, α Ha, J = 6.8 Hz, J = 13.7 Hz), 3.32 (q, 1H, α Hb, J = 8.6 Hz, J = 13.7 Hz, 3.66 (q, 1H, β H, J = 6.8 Hz, J = 8.6 Hz), 3.65 (s, 3H, OCH₃), 3.75 (s, 6H, OCH₃), 3.98 (q, 2H, J = 7.2 Hz), 4.97 (s, 2H, OCE₂Ph), 5.00 (s, 2H, OCH₂Ph), 6.48 - 6.90 (m, 5H, Ar), 7.22 -7.48 (m, 10H, PhCH₂). Mass spec. (70 ev), m/z 556 (H^{+*},4%), 483 (0.7%), 482 (0.4%), 329 (5%), 227 (35%), 257 (18%), 107 (8%), 91 (100%), 65 (16%).

Ethyl 2,3-di(4-benzyloxy-3,5-dimethoxyphenyl) propanoate, 9

It was prepared as described for compound <u>6</u>, using ethyl 4-benzyloxy-3,5-dimethoxyphenylethanoate (1.65g $\tilde{}$ 5 mmol) and 4benzyloxy-3,5-dimethoxyphenyl chloride (1.46g $\tilde{}$ 5 mmol). Yield 0.88g $\tilde{}$ 30%, m.p. 96 - 98°C (cyclohexane:ethyl ethanoate). Calcd. for C₃₅H₃₈O₈: C, 71.65; H, 6.53%. Found: C, 71.73; H, 6.23%. ¹H nmr, (CDCl₃), & ppm 1.11 (t, 3H, CH₃, J = 7.0 Hz), 3.32 (q, 1H, α Ha, J = 6.8 Hz, J = 13.7 Hz), 3.34 (q, 1H, α Hb, J = 8.6 Hz, J = 13.7 Hz), 3.67 (q, 1H, β H, J = 6.8, J = 8.6), 3.73 (s, 6H, OCH₃), 3.77 (s, 6H, OCH₃), 4.10 (q, 2H, OCH₂CH₃, J = 7.0 Hz), 5.00 (s, 2H, OCH₂-Ph), 5.09 (s, 2H, OCH₂Ph), 6.57 - 6.82 (m, 4H, Ar), 7.15 - 7.41 (m, 10H, PhCH₂). Mass spec. (70 ev), m/z 586 (H^{+*}, 3.5%), 513 (0.8%), 329 (4%), 257 (61%), 107 (8%), 91 (100%), 65 (16%).

2-(4-Benzyloxy-3-methoxyphenyl)-3-(4-benzyloxy-3,5-dimethoxyphenyl)-1-propenol, <u>11</u>

It was prepared as described for compound <u>14</u>. Yield 0.37g \sim 79%, m.p. 94 - 96°C (Skelly B:ethyl ethanoate). Calcd. for

 $C_{32}H_{34}O_6$: C, 74.69; H, 6.66%. Found: C, 74.58; H, 6.79%. ¹H nmr, (CDCl₃), δ ppm 1.45 (s, broad, 1H, OH), 2.68 - 2.90 (m, 3H, α H and β H), 3.57 - 3.70 (m, 2H, γ H, overlapped vith OCH₃), 3.67 (s, 6H, OCH₃), 3.80 (s, 3H, OCH₃), 4.98 (s, 2H, OCH₂Ph), 5.11 (s, 2H, OCH₂Ph), 6.47 - 6.80 (m, 5H, Ar), 7.16 - 7.41 (m, 10H, Ph proton). Mass spec. (70 ev), m/z 514 (M^{+*}, 14%), 483 (0.6%), 257 (56%), 240 (6%), 149 (7%), 107 (10%), 91 (100%), 65 (25%). 3-(4-benzyloxy-3-methoxyphenyl-1-propanol, 12

It was prepared according to the method as described for compound <u>14</u>. Yield ~ 78%, m.p. 98 - 100° C (Skelly B:ethyl ethanoate). Calcd. for $C_{32}H_{34}O_6$: C, 74.69; H, 6.66%. Found: C, 74.78; H, 6.50%. ¹H nmr, (CDCl₃), δ ppm 1.50 (s, broad, 1H, OH), 2.69 - 2.91 (m, 3H, α H and β H), 3.57 - 3.70 (m, 2H, γ H overlapped with OCH₃), 3.66 (s, 3H, OCH₃), 3.76 (s, 6H, OCH₃), 5.00 (s, 2H, OCH₂Ph), 5.03 (s, 2H, OCH₂Ph), 6.50 - 6.80 (m, 5H, Ar), 7.20 -7.50 (m, 10H, PhCH₂). Mass spec. (70 ev), m/z 514 (H^{+*}, 13%), 287 (32%), 227 (33%), 270 (7%), 179 (5%), 107 (10%), 91 (100%), 65 (25%).

2,3-Di(4-benzyloxy-3,5-dimethoxyphenyl)-1-propanol, 13

It was prepared as described for the compound <u>14</u>. Yield 78%. Calcd. for $C_{33}H_{36}O_7$: C, 72.77; H, 6.66%. Found: C, 72.50; H, 6.27%. ¹H nmr, (CDCl₃), δ ppm 1.49 (s, broad, 1H, OH), 2.68 - 2.78 (m, 3H, α H and β H), 3.59 - 3.68 (m, 2H, γ H, overlapped with OCH₃), 3.65 (s, 6H, OCH₃), 3.71 (s, 6H, OCH₃), 4.98 (s, 2H, OCH₂Ph), 5.00 (s, 2H, OCH₂Ph), 6.46 - 6.69 (m, 4H, Ar), 7.15 - 7.34 (m, 10H, PhCH₂). Mass spec. (70 ev), m/z 544 (H⁺, 11%), 287 (33%), 257 (34%), 270 (5%), 179 (6%), 107 (10%), 91 (100%), 65 (25%).

2-(4-Hydroxy-3-methoxyphenyl)-3-(4-hydroxy-3,5-dimethoxyphenyl)-1-propenol, 15

It was prepared according to the method described for compound <u>14</u>. Calcd. for $C_{18}H_{22}O_6$: C, 64.66; H, 6.63%. Found: C, 64.86; H, 6.49%. ¹H nmr (CDCl₃, D₂O), δ ppm 2.69 - 2.89 (m, 3H, α H and β H), 3.60 - 3.69 (m, 2H, γ H, overlapped with OCH₃),

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3.68 (s, 6H, OCH₃), 3.75 (s, 3H, OCH₃), 6.49 - 7.50 (m, 5H, Ar). Mass spec. (70 ev), m/z 334 (H^{+,}, 11%), 150 (37%), 167 (100%). <u>2-(4-Hydroxy-3,5-dimethoxy)-3-(4-hydroxy-3-methoxyphenyl)-1-</u> propanol, <u>16</u>

It was prepared according to the method described for compound <u>14</u>. Calcd. for $C_{18}H_{22}O_6$: C, 64.66; H, 6.63%. Found: C, 64.91; H 6.62%. ¹H nmr (CDCl₃, D₂O), δ ppm 2.67 - 2.88 (m, 3H, α H and β H), 3.59 - 3.68 (m, 2H, γ H overlapped with OCH₃), 3.70 (s, 3H, OCH₃), 3.75 (s, 6H, OCH₃), 6.50 - 7.46 (m, 5H, Ar). Mass spec. (70 ev), m/z 334 (H⁺, 13%), 180 (38%), 197 (100%), 137 (85%).

2,3-Di-(4-hydroxy-3,5-dimethoxyphenyl)-1-propanol, 17

It was prepared according to the method described for compound <u>14</u>. Calcd. for $C_{19}H_{24}O_7$: C, 62.63; H, 6.64%. Found: C, 62.92; H, 6.29%. ¹H nmr (CDCl₃, D₂O), δ ppm 2.68 - 2.77 (m, 3H, α H and β H), 3.58 - 3.69 (m, 2H, γ H overlapped with OCH₃), 3.66 (s, 6H, OCH₃), 3.74 (s, 6H, OCH₃), 6.50 - 6.88 (d, 4H, Ar). Mass spec. (70 ev), m/z 364 (H^{+*}, 10%), 180 (39%), 197 (100%), 167 (67%).

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REFERENCES

- K. Sudo, D.J. Mullord and J.M. Pepper, Can. J. Chem., <u>59</u>, 1028 (1981).
- A. Bugas and C. Dufour, Ann. Pharm. France, <u>17</u>, 453 (1959). Chem. Abs. <u>54</u>, 6623e (1960).
- T. Kametani, S. Kano and T. Kikuchi, Yakugaku Zasshi, <u>86</u>, 423 (1966). Chem. Abs. <u>65</u>, 5438 (1966).
- K. Schofield, R.S. Ward and A.M. Chowdhury, J. Chem. Soc. C., 2834 (1971).

- T. Kametani, H. Yagi, K. Kavamura and T. Kohno, Chem. Pharm. Bull., <u>18</u>, 645 (1970).
- 6. I.T. Strukov, J. Gen. Chem. U.S.S.R., <u>31</u>, 2528 (1961).
- J.H. Short, D.A. Dunnigan and C.V. Ours, Tetrahedron, <u>29</u>, 1931 (1973).
- A.J. Nonni and C.W. Dence, J. Wood Chem. Technol., <u>2</u>, 161 (1982).
- J. Kametani and J. Serizava, J. Pharm. Soc. Japan, <u>72</u>, 1084 - 1087 (1952).
- J.A. King and P.H. McMillan, J. Am. Chem. Soc., <u>68</u>, 2335 (1946).
- 11. M.W. Rathke and A. Lindert, J. Am. Chem. Soc., 93, 9 (1971).